A Level Chemistry

Why study Chemistry?

To gain fundamental understanding of the processes and phenomena that takes up all materials and is the foundations of life itself. Advance in both Physics and Biology are of teintertwined with, and reliant upon, chemical ideas and knowledge. Hum will benefit greatly from the new materials and medicines which inevitably will come from our advancing awareness of the fundamental nature of mater. This course will help students to lay the groundwork for further study incourses such as chemistry, medicine and pharmacy.

What is the course structure?

Students will study three areas of Chemistry: Inorganiog nic and Physical. Overall, at least 15% of the marks all A-level Chemistry courses will require assessment of practical skills. Sturds will sit the A-level exams at the end of their A-level course.

A Level Examinations

In the first year, students explore the fundamentalipeciples that form the basis of chemistry such as atomic structure, bonding, periodicit and an introduction to organic chemistry. Students also look at the applications of these principals. In the second year, students evelop further the concepts and principles introduced through topics including: equilibria, polymers, aromatic chemistry, thermodynamics, energetic chemistry and inorganic chemistry.

Paper 1

What's assessed

- € Relevant physical chemistry topics (Atomic sturce, Amount of substances, Bonding, Energetics, Chemical equilibria and Le Chatelier•s priplei sections, Oxidation, reduction and redox equations, Thermodynamics, Equilibrium constakc for homogeneous systems and Acids and bases)
- € Inorganic chemistry (Periodicity, Group 2, the time earth metals, Group 7, the halogens, Properties of period 3 elements and their oxide ransition metals and Reactions of ions in aqueous solution)
- € Relevant practical skills

Assessed

€ Written exam: 2 hours

€ 105 marks € 35% of A-level

Questions

105 marks of short and long answer questions

Paper 2

What's assessed

€ Relevant physical chemistry topics (Amount dostances, Bonding, Energetics, Kinetics, Chemical equilibria and Le Chatelier•s primote sections and Rate equations)

€ Organic chemistry (Introduction to organic chemistAlkanes, Halogenoalkaes, Alkenes, Alcohols, Organic analysis, Optical isomers, Aldehydes and kets) Carboxylic acids and derivatives, Aromatic chemistry, Amines, Polymers, Amino acids, proteand DNA, Organic synthesis, NMR spectroscopy and Chromatography)

€ Relevant practical skills